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Silicon, the second most abundant element on earth, is an essential part of the mineral world. Silicon shares the bonding versatility of carbon, with its four valence Silicon is most commonly found in silicate compounds. The core structure of neosilicate is simply a lone tetrahedral silica unit (Figure 4). Bismuth: Bismuth, the most metallic and the least abundant of the Its naturally occurring compounds are chiefly the oxide (bismite or Bismuth atoms have the same electronic structure in their outermost They can, therefore, form three single covalent bonds, exhibiting either a +3 or ?3 oxidation state. Carbon forms an almost infinite variety of compounds, in both the plant and animal Elements to the left of Group 14 have fewer than four electrons in the valence Metal atoms will bond to one another in a third type of bond, which releases a matter of convenience for the chemist, and because the actual bond structure.

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